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(54) Title: SANITIZING COMPOSITIONS			
(57) Abstract This invention is a microbicidal and tuberculocidal composition comprising a major portion of carrier and an effective sanitizing amount of octanoic acid, or octanoic acid derivatives, and a sulfur containing compound. Optionally, the invention may also comprise any variety of formulatary ingredient options or application adjuvants. The invention comprises concentrate compositions and methods of sanitizing and disinfecting using the antimicrobial composition of the invention.			

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SANITIZING COMPOSITIONS

Field of the Invention

5 The invention relates to microbicidal compositions for sanitizing inanimate surfaces. More specifically, the invention relates to microbicidal compositions which include an octanoic carboxylic acid and a sulfur containing compound as an antimicrobial agent. The
10 composition is preferably safe for incidental human contact as well as food contact surfaces without requiring a post-sanitizing rinse. The microbicidal compositions of the invention are suitable for dairy farms, food and beverage processing plants, food
15 preparation kitchens, food serving establishments, child-care, nursing-care and hospital-care applications, as well as for general utility in domestic households and institutions.

Background of the Invention

20 The list of microbicidal agents has decreased due to their human toxicity and their detrimental effect on water supplies and the overall environment. Improving analytical capabilities to detect parts-per-billion
25 levels in food, water and in the environment generally have raised important safety concerns about the application and misapplication of these chemicals. These issues have resulted in the banning of some antimicrobials, for example hexachlorophene; the
30 retesting of others for animal toxicity, such as, the quaternary ammonium compounds; and, the increasing scrutiny of microbicidal species such as chlorine or hypochlorites which may form toxic halocarbons in effluent waters.

35 There has been a long felt need for antimicrobial agents which have a high degree of antimicrobial efficacy, and which are preferably safely used around sensitive areas while also posing no environmental incompatibility. Those antimicrobial agents which are

lethal to microorganisms, however, are also toxic in varying degrees to humans and animals in that both higher and lower forms of life share at least some common metabolic pathways. Competitive inhibition, non-
5 competitive inhibition, protein coagulation, oxidative and reductive action, blockage of enzyme systems are thought to be some of the mechanisms involved in the destruction of microorganisms.

Differentiation of antimicrobial "-cidal" or "-static" activity, the definitions which describe the degree of efficacy, and the official laboratory protocols for measuring this efficacy are important considerations for understanding the relevance of antimicrobial agents and compositions. Antimicrobial
15 compositions may effect two kinds of microbial cell damage. The first is a truly lethal, irreversible action resulting in complete microbial cell destruction or incapacitation. The second type of cell damage is reversible, such that if the organism is rendered free
20 of the agent, it can again multiply. The former is termed bactericidal and the latter, bacteriostatic. Sanitizers, disinfectants and tuberculocidal agents are, by definition, agents which provide bactericidal activity. In contrast, a preservative is generally
25 described as inhibitory or bacteriostatic.

A sanitizer is an agent that reduces the number of bacterial contaminants to safe levels as judged by public health requirements. Practically, a sanitizer must result in 99.999% reduction (5 log order reduction)
30 for given organisms as defined by Germicidal and Detergent Sanitizing Action of Disinfectants, Official Methods of Analysis of the Association of Official Analytical Chemists ("A.O.A.C."), paragraph 960.09 and applicable sections, 15th Edition, 1990 (EPA Guideline
35 91-2).

A disinfectant is an agent that kills all vegetative cells including most recognized pathogenic microorganisms. As such, it must pass a more stringent

bactericidal test; the A.O.A.C. Use Dilution Methods,
Official Methods of Analysis of the Association of
Official Analytical Chemists, paragraph 955.14 and
applicable sections, 15th Edition, 1990 (EPA Guideline
5 91-2).

A tuberculocide is a higher order disinfectant which
is capable of killing all vegetative tuberculosis
bacteria cells. Tuberculocidal activity is determined
by Tuberculocidal Activity of Disinfectants, Official
10 Methods of Analysis of the Association of Official
Analytical Chemists, paragraph 965.12 and applicable
sections, 15th Edition, 1990.

In contrast, a preservative is described as any
agent that generally extends the storage life of
15 perishable products such as food and non-food products
by retarding or preventing deterioration of flavor,
odor, color, texture, appearance, nutritive value, or
safety. One method used for evaluating such materials
is designated Minimum Inhibitory Method Concentration.
20 Another procedure is entitled Zone of Inhibition.
Preservatives, by definition, are therefore inhibitory
substances added to food to prolong or enhance shelf-
life. The principal differences between a preservative
and a sanitizer are two-fold; 1) mode of action, a
25 preservative prevents growth rather than killing
microorganisms; and, 2) exposure time, a preservative
has days to months. In contrast, a sanitizer must
provide 99.999% kill (5 log order) within 30 seconds at
nominal 20°C.

30 Ideally, a sanitizing agent or compound will possess
several important properties in addition to its
microbicidal efficacy. The sanitizer should be no-rinse
after application, and have residual antimicrobial
activity. Residual activity implies a film of
35 sanitizing material which will continue to have
antimicrobial effect if the treated surface is
contaminated by microorganisms during a storage or lag
period. The sanitizer should be odor free to prevent

transfer of undesirable odors onto contact surfaces or articles with which it otherwise comes into contact. The sanitizer should be composed of ingredients which will not affect food if incidental contact or
5 contamination occurs, nor affect humans should incidental ingestion result. In addition, the sanitizer should be composed of naturally occurring or innocuous ingredients, which are chemically compatible with the environment and cause no concern for toxic residues in
10 downstream water.

Previously, certain compositions have been recognized as effective in providing sanitizing, disinfecting, and preservative effects. For example, U.S. Patent No. 4,404,040 to Wang discloses the
15 sanitizing properties of short chain fatty acids formulated with an ionic hydrotrope-solubilizer and compatible acids. U.S. Patent No. 4,647,458 to Ueno et al, discloses bactericidal compositions comprising a large concentration of ethyl alcohol, an organic acid,
20 and an inorganic acid.

Moreover, U.S. Patent No. 3,915,633 to Ramachandran, discloses a prewash composition for treating fabrics which includes an organic acid such as citric acid and either a nonionic or an anionic surfactant. U.S. Patent
25 No. 3,867,300 to Karabinos, discloses bactericidal compositions presumably for controlling the spread of nosocomial infections in hospitals consisting of an aliphatic monocarboxylic acids, and nonionic surfactants. U.K. Patent Application GB 2,103,089A to
30 Kimberly Clark discloses the use of carboxylic acids as virucides. U.S. Patent No. 4,715,980 to Lopes et al, discloses an antimicrobial concentrate composition containing a dicarboxylic acid, a solubilizer, an acid, and a diluent. U.S. Patent No. 3,650,965 to Cantor et
35 al, discloses clean-in-place detergent solutions for treating milk and food processing equipment based on two different nonionic surfactants.

U.S. Patent No. 4,002,775 to Kabara discloses the

use of mono-esters of twelve carbon aliphatic fatty acids and polyols. European Patent Application No. 87303488 to Kabara discloses antimicrobial preservative compositions of glycerol mono esters, preferably monolaurin and fatty acids. However, similar to Wang and Ueno et al, the disclosure in these publications is not specific to C₈ acids and further does not discuss the antimicrobial activity of these acids in conjunction with their use with certain adjuvants.

Currently, products used for sanitizing operations include strong oxidizing agents such as peracetic acid, iodophors, sodium hypochlorite and related n-chloro compounds such as chloro isocyanurates, quaternary ammonium compounds and the like. While these are no rinse sanitizers, they are not ideal for one reason or another.

Peracetic acid, iodophors and chlorine based sanitizers are either decomposed or lost by evaporation when a film of sanitizer is left on the contact surface and allowed to dry. Thus no residual activity remains on the intended surface. Residual activity is necessary to provide continued antimicrobial effect if the surface is contaminated by microorganisms during storage.

Quaternary ammonium compounds (QAC) have an excellent residual quality as they are stable and increase in concentration as the solvent (water) evaporates. Unfortunately, for many uses, this residue may carry into sensitive areas which do not tolerate QAC residues. For example, trace amounts of QAC in substances such as milk, inhibits the starter culture which produces lactic acid and flavor resulting in the curdling of milk protein.

Acid based sanitizers often contain foam control agents or surfactant couplers which are also incompatible in sensitive areas. Moreover, carboxylic acid based sanitizers often have undesirable organoleptic properties exemplified by a "goat-like" odor. The longer chain fatty acids have limited

solubilities in water and require thorough rinsing with potable water before contact of the sanitized surface to avoid imparting odors or flavors to articles contacting the surface.

5 While all these compositions are excellent sanitizers, many of their ingredients are not applicable or otherwise compatible with contact sensitive surfaces. Consequently, these current, commercially successful products have not been designed for user safety,
10 misapplication or environmental compatibility. Thus a sanitizing agent which specifically addresses these issues would possess utility and uniqueness not found in heretofore described sanitizers.

15 Summary of the Invention

The invention is based on the surprising discovery of an antimicrobial composition which is capable of providing sanitizing and disinfecting antimicrobial efficacy as well as tuberculocidal activity. We have
20 found that octanoic acid and when combined with various sulphur containing compounds have an unexpected level of antimicrobial properties in comparison to other antimicrobial compositions.

The composition of the invention generally comprises
25 a carrier and an antimicrobial agent of octanoic carboxylic acid and a sulfur compound. Optionally, the invention may also contain a variety of formulatory or application adjuvants. The invention also comprises concentrate compositions and methods of sanitizing and
30 disinfecting using the antimicrobial composition of the invention.

The claimed composition eliminates the potential for recontamination of sanitized surfaces by potable water which may be safe to drink but may contain
35 microorganisms. This is particularly important in environments such as, for example, where there is a delay between sanitizing operation and use of food preparation equipment. In cases where equipment remains

wet between uses, contaminating organisms may grow freely. Airborne contamination may also be retarded by the invention by retention of compositional residue on surfaces. Especially in the presence of moisture, this
5 residue will continue its antimicrobial action. When residual amounts of the invention are retained on the surface of application, continued sanitizing action will occur in the face of exposure to contaminating splash and spray.

10 The invention is also applicable to closed systems such as pipelines and holding tanks which may be difficult to completely drain. When used, the invention will continue to effectively destroy any microorganisms which might be present without creating risk of harmful
15 food contamination or environmental contamination.

Detailed Description of the Invention

The invention comprises a composition capable of imparting sanitizing and disinfecting antimicrobial
20 efficacy as well as tuberculocidal activity. The composition may also comprise an acidulant along with any variety of other formulatortory or application adjuvants.

The invention also comprises concentrate and use
25 dilution formulations which may take the form of liquid solutions, gels, as well as impregnated sponges, towelettes, aerosol and pump sprays or solids. The invention further comprises methods of sanitizing and disinfecting using the composition of the invention.

30 I. Antimicrobial Agent

The composition of the invention generally comprises an antimicrobial agent.

The invention is based on a discovery that a specific carboxylic acid, octanoic acid when combined
35 with a sulfur containing compound, surprisingly provides extraordinary sanitizing, if not tuberculocidal, disinfecting, antimicrobial efficacy.

Generally, the antimicrobial agent of the invention

functions to sanitize or disinfect the intended surface of application. Further, the composition of the invention also provides tuberculocidal activity. The antimicrobial agent of the invention is intended to
5 provide tuberculocidal, sanitizing or disinfecting antimicrobial activity upon application to the intended surface, leaving a residue which upon contact with foodstuffs will not contaminate or otherwise preclude ingestion of the prepared food.

10 Generally, the composition of the invention is applicable to all food collection, process, preparation and serving environments and facilities as well as other contact sensitive areas such as day and child care facilities, nursing homes and other health care
15 facilities, and domestic households.

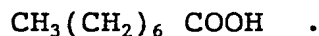
Thus, a sanitizer and disinfectant which is excellent microbicidally, does not require a post-sanitizer rinse, imparts no off-flavor or odor to food, possess residual activity, and minimizes the potential
20 for acute and chronic human toxicity and environmental contamination fulfills a need not currently met by presently available sanitizers. The antimicrobial agent of the invention comprises a carboxylic acid system of octanoic acid and derivatives thereof combined
25 with a sulfur containing compound. Carboxylic acids are characterized by the presence of one or more carboxyl groups which generally have the structure:



Carboxylic acids as a group are usually considered to be relatively weak acids.

Even in view of the weakness of these acids, we have
35 found that one carboxylic acid provides unique antimicrobial efficacy despite this classification. The antimicrobial agent of the invention consists of

octanoic acid as well as, octanoic acid esters, or salts. Octanoic acid also known as caprylic acid, occurs naturally as glycerides and may generally be derived by saponification and subsequent distillation of coconut oil. Octanoic acid is generally an oily liquid having a boiling point of 239.7°C., a melting point of 16.7°C. and a density of 0.910 (at 20°C.). Octanoic acid is known by the formula:



10 In addition to antimicrobial efficacy resulting from simple octanoic acid, antimicrobial efficacy may also result from octanoic acid esters, or salts. Specifically, the carboxylic acid of the invention may also be derivatized into the form of a carboxylic acid ester, or carboxylic acid salt. Further, as with all 15 carboxylic acids, industrial grades of octanoic acid may also comprise minor proportions of other carboxylic acids as impurities.

Generally, the linear carboxylic acid of the invention may also take the form of a salt formed by 20 reaction with an alkaline substance most commonly from oxides, hydroxides or carbonates of monovalent and divalent metals in Periodic Groups IA and IIA; but, also with basic positive complexes such as the ammonium radical and organic amine moieties. 25

The carboxylic acid of the invention may also comprise an ester derivative of that carboxylic acid. Common ester derivatives of carboxylic acids are those

wherein the hydroxy group is replaced by an alkoxy group which may comprise any number of different alkyl moieties which do not impede the efficacy of the octanoic acid compound.

- 5 The principal types of esters result from reaction with monohydric alcohols, polyhydric alcohols and ethylene or propylene oxide. The most common monohydric alcohols used are the simple alkyl alcohols such as methyl, ethyl, propyl, isopropyl, and the like. The
10 most common polyhydric alcohols include polyethylene glycol, glycerol, sorbitol, and certain carbohydrates such as sucrose.

Accordingly, the octanoic carboxylic acid of the invention may comprise any number of acid salts, acid
15 esters, and the like. Preferably, the compound used in the invention is octanoic acid or an octanoic acid salt or an octanoic acid ester.

Generally, depending on whether the composition is a use dilution or concentrate formulation, octanoic acid
20 may be present in concentrations ranging generally from about 0.01 wt-% to about 45 wt-% preferably from about 0.03 wt-% to about 40 wt-%, and most preferably from about 0.05 wt-% to about 35 wt-%.

The concentration figures detailed above for
25 octanoic acid are presented as guidelines. Actual concentrations vary depending upon the carrier used in the formulation, whether aqueous, organic, inorganic or mixtures thereof; the overall nature of the formulation,

whether neat solution, liquid concentrate, or aerosol, dispersion, emulsion, gel, or solid; the delivery and application method; and, the compositional adjustments necessary for physical and chemical stability during
5 storage or use in adverse environments.

Additionally, the antimicrobial agent of the invention also comprises a compound containing sulfur. Sulfur compounds and especially compounds such as sulfonates, and sulfates, among others, provide a
10 tuberculocidal, sanitizing and disinfecting antimicrobial character when combined with octanoic acid and derivatives thereof. Further, these sulfur compounds may also function to increase acidity, as well as provide surface activity and coupling within the
15 composition. Generally, this agent may comprise any compound, surfactant, polymer, or mixture thereof containing sulfur. Preferably the sulfur compound comprises an organic sulfonic acid moiety or sulfuric acid ester to provide antimicrobial efficacy, acidity,
20 and surface activity.

Generally, the sulfur compound may comprise an aliphatic, aromatic, or alicyclic structure and derivitized combinations thereof which have been subjected to sulfonation, or sulfation reactions. In
25 sulfonation, a new C-S bond is created and a SO_3^- group is introduced into an organic molecule to provide a derivative with a C- SO_3^- linkage, a grouping known as a sulfonate which may remain protonated (sulfonic acid),

or be neutralized with base (sulfonic acid salts). Sulfation results from any process of introducing an SO_3 group into an organic compound by forming a C-O or O-S of the C-O-S bond sequence. The reaction product, a sulfate, exhibits the characteristic $-\text{C}-\text{O}-\text{SO}_3^-$ configuration.

Generally, in the context of the claimed invention, the acid form of a sulfonated or sulfated compound or polymer is preferred. Compounds which may be sulfonated or sulfated for use in accordance with the invention include the acid and various salt derivatives of sulfonated paraffins, sulfonated olefins, sulfonated lignins, sulfonated mono and polycarboxylic acids and alcohol esters of these acids; and, sulfonated alicyclic, aromatic, and alkylaryl moieties; also, the acid and corresponding salt compounds of sulfated alcohols and ether alcohols; sulfated glycerol esters of fatty acids; and products obtained by sulfation of saturated, unsaturated and hydroxy fatty acids and natural fats and oils containing their glycerides, as well as monohydric and polyhydric alcohol esters of these acids, among others.

One preferred class of compounds are alkyl-aryl sulfonates such as alkyl benzene sulfonates.

Specifically preferred compounds are aromatic sulfonate compounds such as alkyl benzene sulfonates, decanoic benzene sulfonates, dodecanoic or dodecyl

benzene sulfonates, tetradecanoic benzene sulfonates, and hexadecanoic benzene sulfonates, and mixtures thereof. These compounds may also be used in their acid form as sulfonic acid compounds.

- 5 The most preferred sulfur compound has been found to be dodecyl benzene sulfonic acid as it is a very strong acid affecting protonation of weak fatty acids such as octanoic acid, and is of itself a microbicide as well as a good surfactant.
- 10 Generally, depending on whether the composition is a used dilution or concentrate formulation, octanoic acid may be present in concentrations ranging generally from about 0.01 wt-% to 45 wt-%, preferably from about 0.03 wt-% to 40 wt-%, and most preferably from about 0.05 wt-
- 15 % to 35 wt-%.

II. Carrier

- The antimicrobial composition of the invention also comprises a carrier. The carrier within this composition functions to transport the antimicrobial
- 20 agents to the intended surface of application and define the forms of the composition whether liquid, semi-solid such as a gel, or solid. Moreover, depending upon the nature of the carrier, this constituent may be used to maintain the antimicrobial agent on the intended surface
- 25 for an extended period of time in the form of a film or residue after application. Keeping these functions in mind, the carriers useful in the invention should preferably maintain and not obscure the efficacy of the

antimicrobial agent.

The composition of the invention may take the form of a neat solution or liquid concentrate, dispersion, emulsion, aerosol, gel, or solid. The invention may
5 also take the form of a liquid impregnated sponge or towelette where the carrier comprises, in addition to a liquid, a chemically inert carrier such as a fabric or sponge. Accordingly, the choice of any carrier useful in the invention will depend somewhat on the intended
10 form and intended use application of the final composition. If the invention takes the form of a solution, dispersion, gel, emulsion, aerosol, or solid, useful carriers include water or aqueous systems as well as organic or inorganic based carriers, or mixtures
15 thereof.

Organics which have been found especially useful include simple alkyl alcohols such as ethanol, isopropanol, n-propanol and the like. Polyols are also useful carriers in accordance with the invention,
20 including propylene glycol, polyethylene glycol, glycerol, sorbitol and the like. Any of these compounds may be used singly or in combination with another organic or inorganic carrier or, in combination with water, or in mixtures thereof.

25 If organic, the carrier may also comprise any number of surfactants or surfactant combinations. Surface active agents which have been found as useful carrier in accordance with the invention include anionic and

- nonionic agents such as, for example, propylene glycol esters, glycerol esters, polyoxyethylene glycerol esters, polyglycerol esters, sorbitan esters, polyoxyethylene sorbitan esters, sucrose esters,
- 5 polyethylene glycol esters, polyoxyethylene-polyoxypropylene ether adducts, dioctyl sodium succinate, stearyl lactylate, and esters of acetylated, lactylated, citrated, succinylated or diacetyl tartarated glycerides.
- 10 Preferred surfactants include nonionic surfactants having a mixture of polyoxyethylene and polyoxypropylene moieties. Specifically, one nonionic surfactant found to be especially preferred is a polyoxyethylene, polyoxypropylene block copolymer having about 240 to 280
- 15 moles of ethoxylation and about 45-65 moles of propoxylation.
- If the invention is formulated as a solid, the carrier may be selected from any organic or inorganic compound which imparts a solid form and hardness to the
- 20 composition of the invention either by a hot-melt, pour-cast process, by extrusion, or by compression. Typical organic ingredients which may be used in the solid antimicrobial composition of the invention to harden this composition include amides, polyols, and certain
- 25 nonionic and anionic surfactants.

For example, stearic monoethanol amide, stearic diethanol amide and urea have been found to effectively result in the formulation of a hardened product.

Moreover, polyols such as polyethylene glycol, and polyhydric sugar alcohols such as mannitol and the like or mixtures thereof have all been found to impart a hardened but soluble character when combined in the composition of the invention.

Surfactants useful in this invention as a hardening agent and carrier are solid, generally high melting analogs of nonionics and anhydrous metallic salts of anionic surfactants which include alkyl and dialkyl phenol ethoxylates, linear alkyl alcohol ethoxylates, polyalkoxide polymers of ethanolamines, ethylene oxide/propylene oxide block copolymers, polyalkylene oxide block polymers of ethylene diamine, glycerol esters, polyoxyethylene glycerol esters, polyglycerol esters, sorbitan esters, polyoxyethylene sorbitan esters, sucrose esters, polyethylene ethers, dioctyl sodium sulfo succinate, stearyl lactylate, and complex esters such as acetylated, lactylated, citrated, succinylated, and diacetyl tartarated glycerides.

Other compositions which may be used as hardeners within the composition of the invention include sugars, and modified starches or cellulose which have been made water soluble through acid or alkaline treatment processes.

Inorganics which may be used in forming the hardened antimicrobial composition of the invention include salts formed of Periodic Groups IA and IIA metals, as well as ammonium, with the corresponding negative ions or

radicals of mineral acids such as chloride ions, carbonate ions, nitrate ions, phosphate ions, and sulphate ions as well as their respective hydrates, protic salt forms, or in the case of phosphates, the various condensate species.

Generally, any type of carrier capable of solidifying the antimicrobial agent may be used in accordance with the invention. To this end, urea, Pluronic [™] F-108 and polyethylene glycol have been found to be beneficial solidifying agents.

Generally, the carrier comprises a large portion of the composition of the invention. Here again, the carrier concentration and type will depend upon the nature of the composition as a whole, the environment of storage and method of application including the concentration of antimicrobial agent, among other factors. Notably, the carrier should be chosen and used at a concentration which does not inhibit the antimicrobial efficacy of the active in the present composition.

III. Adjuvants

Alternatively, the composition of the invention may also comprise any number of adjuvants. Depending on the benefits provided by the adjuvant, adjuvants may partially or wholly displace the carrier in the composition. Generally, in accordance with the invention, there may be included within this composition formulatory adjuvants or adjuvants which assist in the

application of the invention with respect to performance, form, aesthetics, and stability when stored or used within adverse conditions.

Formulatory adjuvants include coupling agents, solubilizers, or hydrotropes used to maintain the storage stability of the present composition as well as solubilizing the antimicrobial agent of the invention.

This function may be accomplished exclusively by the carrier whether aqueous, organic, inorganic or a mixture thereof. However, in situations which require formulation of a concentrated antimicrobial system, an additional organic agent may be introduced into the system to facilitate solubilization of the antimicrobial agent.

To this end, any number of organic coupling agents may be used including monofunctional and polyfunctional alcohols. Those coupling agents which have been found most useful include linear alkyl alcohols such as, for example, ethanol, isopropanol, and the like.

Polyfunctional organic alcohols include glycerol, hexylene glycol, polyethylene glycol, propylene glycol, sorbitol and the like. Generally, depending on whether the composition is in the form of a concentrate or use dilution formulation, the concentration of these adjuvant compounds, when used in these capacities, ranges from about 0 wt-% to about 99 wt-%, preferably from about 0.1 wt-% to about 97 wt-%, and most preferably from about 0.15 wt-% to about 95 wt-%.

The invention may also comprise one or more acidulants useful in lowering the pH of the present composition. Acidulants useful in the present composition include lactic acid, phosphoric acid, 5 sulfuric acid, sulfamic acid, adipic acid, tartaric acid, succinic acid, acetic acid, propionic acid, citric acid, malic acid, or mixtures thereof. Further it has been found that a use dilution solution pH ranging from about 1.3 to 4, preferably from about 1.4 to 3, and most 10 preferably from about 1.5 to 2.5 provide the most desirable antimicrobial efficacy.

The composition of the invention may also comprise surface tension altering constituents such as various anionic and nonionic surfactants. These surfactants may 15 be used to maintain constituents in solution over various temperature gradients as well as altering the wettability and cleaning capabilities of the composition of the invention to any variety of surfaces. Any number of surfactants or combinations thereof may be used in 20 accordance with the invention.

The surface active agents which have been found useful in accordance with the invention include anionic and nonionic agents including, for example, propylene glycol esters, glycerol esters, polyoxyethylene glycerol 25 esters, polyglycerol esters, sorbitan esters, polyoxyethylene sorbitan esters, sucrose esters, polyethylene glycol esters, polyoxyethylene-polyoxypropylene ether adducts, dioctyl sodium

succinate, stearyl lactylate, and complex esters such as acetylated, lactylated, citrated, succinylated, or diacetyl tartarated glycerides.

One class of surfactants which has been found
5 especially useful in formulating the various embodiments of the present composition includes nonionic surfactants which have a mixture of hydrophilic and hydrophobic character. Generally, a mixture of hydrophilic and hydrophobic character in the surfactants has been found
10 particularly useful in accordance with the invention and is created by the presence of polyoxyethylene and polyoxypropylene moieties.

Nonionic surfactants which are especially useful include those surfactants having about 5 - 300 moles of
15 ethoxylation and about 10 - 80 of propoxylation. One surfactant which has been found most useful is Pluronic™ F-108 which is a nonionic surfactant generally defined as a polyoxyethylene, polyoxypropylene block copolymer having about 240 to 280 moles of ethoxylation and about
20 45 to 65 moles of propoxylation, sold by BASF-Wyandotte Company Inc. We have found that BASF-Wyandotte Company Inc. Pluronic F-108 is useful for formulating solid and concentrates, and Pluronic L-44, (having about 5 to 15 moles of EO and 10 to 30 moles of PO), is useful for
25 formulating liquid concentrates.

Surface tension altering constituents of the invention may be used in the present composition, regardless of form or application, depending on whether

the composition is a concentrate or use dilution formulation, in concentrations ranging from about 0 wt-% to 60 wt-%, preferably from about 0.01 wt-% to 50 wt-%, and most preferably from about 0.02 wt-% to 40 wt-%
5 depending on whether the surfactant is present for wetting, detergency, or coupling.

Here again, the concentration and type of surfactant used should not inhibit the antimicrobial action of the active within the invention. The concentration of
10 surfactant adjuvant may also vary depending upon the nature of the formulatory composition as a whole, the concentration of antimicrobial agent, as well as the storage environment and method of application among other factors.

15 As the invention may take the form of a spray, either pump or aerosol, adjuvants which may be used with the carrier in the invention include propellants. Any number of propellants may be used including n-butane, isobutane and propane, among others. The concentration
20 of propellant will generally range from about 3 wt-% to about 25 wt-%, preferably from about 4 wt-% to about 20 wt-%, and most preferably from about 5 to about 15 wt-%.

The composition of the invention may also comprise adjuvants which facilitate the application of this
25 composition through various vehicles. Specifically, the composition of the invention is useful as an antimicrobial agent in hand creams, sponges, towelettes, hand cleansers, dips, sprays and washes among other

uses. Accordingly, the composition of the invention may comprise any number of conditioners or emollients, humectants, perfumes, thickeners, opacifiers or particulates, colorants or dyes, cleansers or other agents useful in facilitating the application of the composition of the invention to its intended application.

Table 1 provides a general directory of guideline concentrations for the various compositional forms of the invention.

TABLE 1
USE-DILUTION CONCENTRATION RANGES (wt-%)

	<u>USEFUL</u>	<u>PREFERRED</u>	<u>MOST PREFERRED</u>
ANTIMICROBIAL AGENT	0.02-1.0	0.06-0.7	0.1-0.4
OCTANOIC ACID	0.01-0.5	0.03-0.35	0.05-0.2
SULFUR COMPOUND	0.01-0.5	0.03-0.35	0.05-0.2
CARRIER	54.98-99.98	64.94-99.84	74.9-99.75
ADJUVANTS	0-45	0.1-35	0.15-25
pH	1.3-4	1.4-3	1.5-2.5

LIQUID CONCENTRATE RANGES (wt-%)

	<u>USEFUL</u>	<u>PREFERRED</u>	<u>MOST PREFERRED</u>
ANTIMICROBIAL AGENT	1-90	3-80	5-70
OCTANOIC ACID	0.5-45	1-40	1.5-35
SULFUR COMPOUND	0.5-45	1-40	1.5-35
CARRIER	0-99	0-95	0-91
ADJUVANTS	0-99	2-97	4-95
pH (USE-DILUTION)	1.3-4	1.4-3	1.5-2.5

SOLID CONCENTRATE RANGES (wt-%)

	<u>USEFUL</u>	<u>PREFERRED</u>	<u>MOST PREFERRED</u>
ANTIMICROBIAL AGENT	1-60	2-50	3-40
OCTANOIC ACID	0.5-30	1-25	1.5-20
SULFUR COMPOUND	0.5-30	1-25	1.5-20
CARRIER	40-99	48-96	56-93
ADJUVANT	0-54	2-50	4-41
pH (USE-DILUTION)	1.3-4	1.4-43	1.5-2.5

GEL COMPOSITION RANGES (wt-%)

	<u>USEFUL</u>	<u>PREFERRED</u>	<u>MOST PREFERRED</u>
ANTIMICROBIAL AGENT	1-50	2-40	3-30
OCTANOIC ACID	0.5-25	1-20	1.5-15
SULFUR COMPOUND	0.5-25	1-20	1.5-15
CARRIER	30-94	38-91	47-88
ADJUVANTS	5-70	7-60	9-50
pH (USE DILUTION)	1.3-4	1.4-3	1.5-2.5

The concentrations provided above generally reflect a ratio of octanoic acid to the sulfur compound of about 1:1. This ratio may range from about 1:0.5 to 10, and preferably about 1:0.5 to 2.

In use we have found that a dilution rate which results in an active concentration of ranging from about 500 ppm to 1500 ppm, preferably about 750 ppm to 1250 ppm, and most preferably 900 ppm to 1100 ppm of each of octanoic acid and sulfur containing compounds has been found useful.

The liquid concentrate may comprise water in the form of carrier ranging from about 0 wt-% to 70 wt-%, preferably from about 15 wt-% to 70 wt-%, most preferably from about 30 wt-% to 70 wt-% as a percentage of the total composition. The gel concentrate may comprise water in the form of carrier ranging from about 0 wt-% to 80 wt-%, preferably from about 15 wt-% to 60 wt-%, and most preferably about 25 wt-% to 40 wt-% as a percentage of the total composition.

WORKING EXAMPLES

Following below are formulatory, stability, application and microbiological working examples using the composition of the invention. While the invention is exemplified by the working examples, it is not limited to the examples shown hereinafter.

WORKING EXAMPLES

1 - 40

Formulatory working examples, working examples 1-40, were prepared by combining the antimicrobial of the invention with various constituents to show compatibility as well as antimicrobial efficacy.

Generally, nonionic coupling agents were thought not to be compatible with various fatty acid compounds such as octanoic acid. Contrary to this general statement, the working examples of Table 2 show that octanoic acid when combined into the composition of the invention are compatible with nonionics such as Pluronic™ F-108 (manufactured by BASF/Wyandotte); (all concentrations are in wt-%). This unexpected compatibility, exceeding 1 wt-% nonionic in use dilution, is important in that coupling agents may be used to stabilize the fatty acid against phase separation at extreme temperature. This is especially relevant when a concentrated sanitizer or disinfectant is desired. Moreover, this level of nonionic surfactant was shown to not affect the antimicrobial efficacy of the composition (see Table 3).

TABLE 2
(wt-%)

COMPONENT	1	2	3	4	5	6	7	8
Octanoic Acid	32.00	32.00	28.57	29.63	30.19	30.32	32.00	30.48
LAS* (97% w/v)	20.00	20.00	17.86	18.52	18.87	18.95	20.00	19.05
Distilled Water	20.00	34.00	33.93	39.81	40.57	40.73	43.00	40.95
Hexylene Glycol	28.00	14.00	6.25	7.41	5.66			4.76
Nonylphenol Ethoxylate (9.5 moles EO)			13.39					
Pluronic™ F-108**				4.63				
Pluronic™ F-38**					4.72			
Pluronic™ L-44**						10.00		
Phosphate ester (acid) of a C ₁₀₋₁₄ alcohol ethoxylate (60 moles EO)								
Alcohol ethoxylate (C ₁₀₋₁₄ 20 moles EO)							5.00	4.76

(* Linear alkyl sulfonate)

(** Nonionic surfactants sold by BASF/Wyandotte)

TABLE 2 (cont.)

COMPONENT	9	10	11	12	13	14	15	16
Octanoic Acid	31.37	25.60	24.33	22.82	24.38	24.38	23.46	24.38
LAS* (97% w/v)	19.61	10.00	9.50	8.92	9.52	9.52	27.49	9.52
Distilled Water	42.16	49.40	46.96	44.04	47.05	47.05	36.11	47.05
Hexylene Glycol			14.46				12.94	
Nonylphenol Ethoxylate (9.5 moles EO)					4.76			
Nonylphenol Ethoxylate (15 moles EO)	6.86							
Pluronic™ F-108		5.00	4.75	4.46		4.76		4.7
Sodium Lauryl Sulfate		5.00						
Propylene Glycol		5.00		19.76				
Sodium Xylene Sulfonate					14.29			
Urea						14.29		
Lactic Acid								14.2
(* Linear alkyl sulfonate)								

TABLE 2 (cont.)

<u>COMPONENT</u>	17	18	19	20	21	22	23	24
Octanoic Acid	24.38	25.60	25.60	25.60	24.98	26.67	24.38	25.60
LAS* (97% w/v)	19.05	20.00	20.00	20.00	12.20	12.50	19.05	12.00
Distilled Water	42.28	39.40	39.40	39.40	55.01	49.38	42.29	47.40
Hexylene Glycol		10.00	10.00					
Butyl Carbitol					2.93		9.52	10.00 ²⁸
Methyl Carbitol						6.25		
Pluronic™ F-108**		5.00					4.76	5.00
Pluronic™ F-38**			5.00	5.00	4.88	5.20		
Pluronic™ L-44**				10.00				
Lactic Acid	14.29							

(* Linear alkyl sulfonate)

(** Nonionic surfactants sold by BASF/Wyandotte)

TABLE 2 (cont.)

<u>COMPONENT</u>	25	26	27	28	29	30	31	32
Octanoic Acid	26.95	25.60	25.60	25.60	24.62	25.60	25.60	25.60
LAS* (97% w/v)	12.63	20.00	12.00	12.00	19.23	25.00	12.00	20.00
Distilled Water	49.90	41.90	49.90	42.40	41.72	34.40	42.40	44.40
Hexylene Glycol		10.00	10.00	10.00	4.81	5.00	10.00	5.00
Isopropanol (91% w/v)	5.26							
Pluronic™ F-108**	5.26	2.50	2.50					
Pluronic L™-44**				10.00	9.62	10.00		5.00
Ecolab LF071 (BLOCK COPOLYMER (mw 1400))							10.00	

(* Linear alkyl sulfonate)
 (** Nonionic surfactants sold BASF/Wyandotte)

TABLE 2 (cont.)

COMPONENT	33	34	35	36	37	38	39	40
Octanoic Acid	12.80	12.80	12.80	6.40	25.60	25.60	25.60	12.80
LAS* (97% w/v)	12.80	20.00	12.80	12.80	25.00	25.00	25.00	12.80
Distilled Water	64.40	57.20	57.40	67.80	34.40	34.40	33.40	56.40
Hexylene Glycol			12.00	8.00	5.00	5.00	6.00	13.00
Pluronic L TM -44**	10.00	10.00	5.00	5.00	10.00		10.00	5.00 ³⁰
Pluronic TM L-35**						10.00		

(* Linear alkyl sulfonate)

(** Nonionic surfactants sold by BASF/Wyandotte)

TABLE 2 (cont.)

<u>COMPONENT</u>	41	42
Octanoic Acid	6.40	
LAS*(97% w/v)	12.80	25.00
Distilled Water	66.80	59.00
Hexylene Glycol	9.00	6.00
Pluronic L ^m -44**	5.00	10.00

(*Nonionic surfactant sold by BASF/Wyandotte)
(**Linear alkyl sulfonate)

The following results were obtained from the 30 day reading of the tuberculocidal activity¹ test for linear alkyl sulfonate compositions diluted to 500 ppm synthetic hard water.

TABLE 3

<u>Working Examples</u>	<u>Concentration</u>	<u>Proskauer- Beck*</u>	<u>Kirshners*</u>	<u>Middlebrook*</u>
39	1 oz/2 gal	10/10	10/10	10/10
39	2 oz/3 gal	10/10	10/10	10/10
42	1 oz/2 gal	10/10	10/10	10/10
42	2 oz/3 gal	10/10	10/10	10/10
Control 1 (phenol)	1:50	10/10	10/10	10/10
Control 2 (phenol)	1:70	8/10	10/10	9/10

The results indicate tuberculocidal efficacy is being achieved with a ten minute exposure time using either the 1 ounce per 2 gallons or 2 ounces per 3 gallons dilution.

*(# Negative Tubes/# Tubes Tested)

¹ Tuberculocidal Activity Disinfectants, Official Methods of Analysis of Official Analytical Chemists, Paragraph 969.12 and applicable sections, 15 Edition, 1990.

EXAMPLES 43-44

An A.O.A.C. Sterilant Test was performed on the formulations shown in Table 4A against *C. sporogenes* on silk sutures at a temperature of 80°C with a 2.5 minute exposure time. Products were prepared in 100 ppm hard H₂O at concentrations of 3, 4, 5, 6, & 7%. Results are as follows:

TABLE 4A
(wt-%)

<u>COMPONENT</u>	<u>EXAMPLE 43</u> (wt-%)	<u>EXAMPLE 44</u>
Octanoic Acid	32.0	25.6
Pluronic™ F-108**	19.2	
Lactic Acid (88 w/v)	48.8	
Pluronic™ L-44**		10.0
Dodecyl Benzene		25.0
Sulfonic Acid (97% w/v)		33.4
Distilled Water		6.0
Hexylene Glycol		

TABLE 4B

<u>Example</u>	<u>Conc.</u>	<u>Primary Growth Tube</u>	<u>Secondary Tube Growth</u>
43	3%	17/20	16/20
	4%	20/20	19/20
	5%	20/20	20/20
	6%	20/20	20/20
35	7%	20/20	20/20
<hr/>			
44	4%	20/20	20/20
	5%	20/20	20/20
	6%	20/20	20/20

**Pluronics™ are EO/PO block copolymers of BASF/Wyandotte

Table 5 depicts representative formulations for various compositions within the scope of the invention.

TABLE 5
(Wt-%)

<u>COMPONENTS (wt-%)</u>	<u>Aerosol</u>	<u>Hard Surface Wipes</u>	<u>Hand Wipes</u>	<u>Udder Wipe Sanitizing</u>	<u>Udder Prewipe</u>
Deionized Water	72.835	76.60	72.85	69.75	78.75
Ethanol	17.100	18.00	18.00	18.00	14.00
Octanoic Acid	0.143	0.15	0.15	0.15	0.10
Lactic Acid	0.143	0.15	1.00	0.50	0.15
Citric Acid			3.00	3.50	
Propylene Glycol	4.750	5.00	5.00	5.00	5.00
Glycerol USP				3.00	2.00
Pluronic™ F-108	0.029	0.10		0.10	
Propellant A-31*	5.000				
<u>(Isobutane)</u>					

The above specification, examples and data provide a complete description of the manufacture and use of the composition of the invention. Since many embodiments of the invention can be made without departing from the spirit
5 and scope of the invention, the invention resides in the claims hereinafter appended.

WE CLAIM AS OUR INVENTION:

1. An antimicrobial composition comprising a carrier and an effective sanitizing amount of antimicrobial, said antimicrobial comprising a linear C₈ carboxylic acid, or
5 derivatives thereof, and a sulfur containing compound.
2. The composition of claim 1 wherein said linear C₈ carboxylic acid derivative is selected from the group consisting of an alkyl carboxylic acid salt, an alkyl carboxylic acid ester, or mixtures thereof.
- 10 3. The composition of claim 2 wherein said octanoic acid is present in a concentration ranging from about 0.01 wt-% to 0.5 wt-%.
4. The composition of claim 1 wherein said sulfur containing compound is selected from the group consisting
15 of a sulfonate, a sulfate, and mixtures thereof.
5. The composition of claim 1 wherein the sulfur containing compound comprises a sulfonate surfactant, a sulfate surfactant, and mixtures thereof.
6. The composition of claim 1 wherein the sulfur
20 containing compound comprises dodecyl benzene sulfonate.
7. The composition of claim 1 wherein the sulfur containing compound is present in a concentration ranging from about 0.01 wt-% to 0.5 wt-%.
8. The composition of claim 1 wherein said carrier is
25 selected from a group consisting of an aqueous solvent, an organic solvent, and mixtures thereof.

9. The composition of claim 8 wherein said carrier is present in a concentration ranging from about 54 wt-% to 98 wt-%

10. The composition of claim 8 wherein said carrier
5 comprises water.

11. The composition of claim 1 wherein the compositional pH ranges from about 1.3 to 4.

12. The composition of claim 8 wherein said organic solvents are selected from the group consisting of an
10 organic monofunctional alcohol, an organic polyfunctional alcohol, or mixtures thereof.

13. The composition of claim 1 additionally comprising an acidulant, said acidulant selected from the group consisting of an organic acid, an inorganic acid, or
15 mixtures thereof.

14. The composition of claim 13 wherein said acidulant is selected from the group consisting of phosphoric acid, sulfuric acid, sulfamic acid, adipic acid, tartaric acid, succinic acid, acetic acid, fumaric acid, propionic acid,
20 citric acid, malic acid, lactic acid, or mixtures thereof.

15. The composition of claim 1 wherein said antimicrobial composition comprises a solid, said carrier comprising a solidifying agent selected from the group consisting of an organic hardening agent, an inorganic
25 hardening agent, or mixtures thereof.

16. The composition of claim 15, wherein said organic

solidifying agent is selected from the group consisting of urea, alkyl and dialkyl phenol^o ethoxylates, linear alkyl alcohol ethoxylates, polyalkoxide block polymers of ethanalamines, polyethylene glycol, polyoxyethylene-
5 polyoxypropylene polymers or mixtures thereof.

17. The composition of claim 15 wherein said solid is present in a concentration ranging from about 40 wt-% to 99 wt-%.

18. The composition of claim 1, wherein said
10 composition comprises a solid concentrate.

19. The composition of claim 1, wherein said composition comprises a liquid concentrate.

20. The composition of claim 1, wherein said composition comprises a gel concentrate.

15 21. A liquid antimicrobial concentrate composition comprising:

(a) from about 1 wt-% to 90 wt-% antimicrobial agent comprising octanoic acid or derivatives thereof, and a sulfur containing compound;

20 (b) from about 0 wt-% to 60 wt-% of a nonionic surfactant;

(c) an effective amount of acidulant to provide a pH from about 1.3 to 4 upon dilution; and

(d) a carrier.

25 22. The composition of claim 21 wherein said linear C₈ carboxylic acid derivative is selected from the group

consisting of an alkyl carboxylic acid salt, an alkyl carboxylic acid ester, or mixtures thereof.

23. The composition of claim 21 wherein the sulfur containing compound comprises a sulfonate surfactant, a sulfate surfactant, and mixtures thereof.

24. The composition of claim 21 wherein the sulfur containing compound comprises dodecyl benzene sulfonate.

25. The composition of claim 21 wherein said nonionic surfactant comprises an ethylene oxide/propylene oxide copolymer comprising about 5 to 15 moles of ethylene oxide and about 10 to 30 moles of propylene oxide.

26. A solid antimicrobial concentrate composition consisting essentially of:

(a) from about 1 wt-% to 60 wt-% antimicrobial agent consisting of octanoic acid or derivatives thereof, and a sulfur containing compound;

(b) from about 0 wt-% to 60 wt-% of a nonionic surfactant;

(c) an effective amount of acidulant to provide a pH of 1.3 to 4 upon dilution; and

(d) a carrier.

27. The composition of claim 26 wherein said linear C₈ carboxylic acid derivative is selected from the group consisting of an alkyl carboxylic acid salt, an alkyl carboxylic acid ester, or mixtures thereof.

28. The composition of claim 26 wherein the sulfur

containing compound comprises a sulfonate surfactant, a sulfate surfactant, and mixtures thereof.

29. The composition of claim 26 wherein the sulfur containing compound comprises dodecyl benzene sulfonate.

5 30. The composition of claim 26 wherein said nonionic surfactant comprises an ethylene oxide/propylene oxide copolymer comprising about 100 to 150 moles of ethylene oxide and about 40 to 70 moles of propylene oxide.

31. An antimicrobial gel comprising:

10 (a) from about 1 wt-% to 50 wt-% octanoic acid, or derivatives thereof, and a sulfur containing compound;

 (b) from about 5 wt-% to 40 wt-% of a nonionic surfactant;

15 (c) an effective amount of acidulant to provide a pH from about 1.3 to 4 upon dilution; and

 (d) a carrier.

32. The composition of claim 31 wherein said linear C₈ carboxylic acid derivative is selected from the group
20 consisting of an alkyl carboxylic acid salt, an alkyl carboxylic acid ester, or mixtures thereof.

33. The composition of claim 31 wherein the sulfur containing compound comprises a sulfonate surfactant, a sulfate surfactant, and mixtures thereof.

25 34. The composition of claim 31 wherein the sulfur containing compound comprises dodecyl benzene sulfonate.

35. The composition of claim 31 wherein said nonionic surfactant comprises an ethylene oxide/propylene oxide copolymer comprising about 100 to 150 moles of ethylene oxide and about 40 to 70 moles of propylene oxide.

5 36. A liquid concentrate antimicrobial composition comprising:

(a) from about 1 to 90 wt-% antimicrobial agent comprising octanoic acid or derivatives thereof and dodecyl benzene sulfonic acid;

10 (b) from about 0.1 to 90 wt-% hexylene glycol;

(c) from about 0.02 to 40 wt-% nonionic surfactant, said surfactant comprising from about 5 to 15 moles ethylene oxide and from about 10 to 30 moles propylene oxide; and

15 (d) a balance of water.

37. A method of using an antibacterial composition, said method comprising the steps of:

(a) providing an antimicrobial composition comprising a major portion of carrier and an effective sanitizing amount of octanoic acid, or derivatives thereof, and a sulfur containing compound; and

20

(b) applying said composition to the intended surface.

38. The method of claim 22, comprising the step of wiping said composition from said surface wherein said composition results in a noncontaminating residue.

25

39. The method of claim 22 wherein said alkyl carboxylic acid comprises octanoic acid, an octanoic acid salt, an octanoic acid ester, or mixtures thereof.

40. The method of claim 22 wherein said carrier is
5 selected from a group consisting of water, organic solvents, and mixtures thereof.

41. The method of claim 22 wherein the compositional pH ranges from about 1.3 to 4.

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 A01N37/02 //(A01N37/02,41:04,41:02)

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A01N

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	CHEMICAL ABSTRACTS, vol. 109, no. 25, 19 December 1988, Columbus, Ohio, US; abstract no. 228943q, UNITED STATES FOOD AND DRUG ADMINISTRATION 'Indirect food additives; adjuvants, production aids, and sanitizers' see abstract & FED. REGIST., vol.53, no.162, 22 August 1988 pages 31835 - 31837 --- -/--	1-41

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

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"O" document referring to an oral disclosure, use, exhibition or other means

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"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

4 October 1994

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Decorte, D

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	CHEMICAL ABSTRACTS, vol. 111, no. 11, 11 September 1989, Columbus, Ohio, US; abstract no. 95735r, UNITED STATES FOOD AND DRUG ADMINISTRATION 'Indirect food additives; adjuvants, production aids, and sanitizers' see abstract & FED. REGIST., vol.54, no.96, 19 May 1989 pages 21618 - 21622 ---	1-41
X	US,A,3 867 300 (J.V. KARABINOS) 18 February 1975 cited in the application see claim 1 see column 2, line 2 - line 5 ---	1-41
X	EP,A,0 135 898 (INTERMEDICAT) 3 April 1985 see claims 1,4 see page 6, line 12 - line 23 see example 23 ---	1-41
X	DE,A,32 29 097 (SCHÜLKE & MAYR) 9 February 1984 see claims see page 5, last paragraph ---	1-41
X	WO,A,83 00163 (ECONOMICS LABORATORY) 20 January 1983 see claims 1-5,8,9 see page 6, line 14 - line 38 see example 1 & US,A,4 404 040 (ID) cited in the application ---	1-41
X	EP,A,0 147 102 (DIVERSEY CORP.) 3 July 1985 see page 9; table 2 ---	1-41
A	WO,A,92 21238 (ECOLAB) 10 December 1992 see claims -----	1-41

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 94/07738

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US-A-3867300	18-02-75	NONE	
EP-A-0135898	03-04-85	DE-A- 3333443	28-03-85
DE-A-3229097	09-02-84	CA-A- 1244759	15-11-88
WO-A-8300163	20-01-83	US-A- 4404040	13-09-83
		AU-B- 561736	14-05-87
		AU-A- 8769582	02-02-83
		CA-A- 1188188	04-06-85
US-A-4404040	13-09-83	AU-B- 561736	14-05-87
		AU-A- 8769582	02-02-83
		CA-A- 1188188	04-06-85
		WO-A- 8300163	20-01-83
EP-A-0147102	03-07-85	DE-A- 3344328	20-06-85
		AU-A- 3637184	13-06-85
WO-A-9221238	10-12-92	AU-A- 2223592	08-01-93
		CA-A- 2107357	05-12-92
		NZ-A- 239646	27-09-94
		US-A- 5234719	10-08-93